



MAHARAJA RANJIT SINGH PUNJAB TECHNICAL UNIVERSITY BATHINDA-151001 (PUNJAB), INDIA

(A State University Estb. by Govt. of Punjab vide Punjab Act No. 5 of 2015 and Approved u/s 2(f) & 12 (B) of UGC; Member AIU)

Department: **CHEMISTRY**

Program: M.Sc. 2016 onwards

COURSE ARTICULATION MATRIX (STUDY SCHEME: 2016 ONWARDS)

Subject	S Code	Semester	Credit	Duration (Hrs)	L	T	P	COs	Statement	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8
Electronic Spectra & Magnetic Properties of Transition Metal	MCHM1-101	1	4	45	4.0.0			CO1	Interpretation of electronic and magnetic properties.	1		2					
								CO2	Interpretation of molecular orbital diagrams of octahedral and tetrahedral diagrams for various electronic properties.	1		2					
								CO3	Concepts of symmetry and group theory in solving chemical structural problems.	1		2					
								CO4	Use of character tables and application of group theory in spectroscopy. UNIT-I	1		1					

Inorganic Chemistry Lab.-I	Subject	Group Theory	Subject	Subject
MCHM1-104	S Code	MCHM1-158	S Code	S Code
1	Semester	1	Semester	Semester
3	Credit	4	Credit	Credit
	Duration (Hrs)	45	Duration (Hrs)	Duration (Hrs)
0.04	I T P	4.0.0	I T P	I T P
CO1	COs	CO1	COs	CO3
To develop basic understanding of various lab practices including safety measures.	Statement	Symmetry elements and point groups.	Statement	To familiarize polymer characterization with various spectroscopic techniques.
1	PO1	1	PO1	
2	PO2	1	PO2	3
	PO3	2	PO3	
	PO4	2	PO4	3
	PO5	1	PO5	
	PO6	2	PO6	
3	PO7		PO7	
	PO8		PO8	
CO2	COs	CO2	COs	CO4
To synthesize inorganic complexes and their characterization.	Statement	Use of character table in spectroscopy	Statement	To learn molecular weight measurement by osmometry, mass spectrometry and Viscometry.
1	PO1	1	PO1	
2	PO2	2	PO2	
	PO3	2	PO3	
	PO4	2	PO4	
	PO5	1	PO5	
	PO6		PO6	
3	PO7		PO7	
	PO8		PO8	

Subject	Subject	Subject	Subject	Subject	Subject
S Code	S Code	S Code	S Code	S Code	S Code
Semester	Semester	Semester	Semester	Semester	Semester
Credit	Credit	Credit	Credit	Credit	Credit
Duration (Hrs)	Duration (Hrs)	Duration (Hrs)	Duration (Hrs)	Duration (Hrs)	Duration (Hrs)
I T P	I T P	I T P	I T P	I T P	I T P
COs	COs	COs	COs	COs	COs
Statement	Statement	Statement	Statement	Statement	Statement
ORG ANO					
MCH MS1-2	(MCHM1-206)				
4	4	2	4	2	4
45	45	45	45	45	45
4.0.0	4.0.0	4.0.0	4.0.0	4.0.0	4.0.0
CO1	CO1	CO1	CO1	CO1	CO1
Organometallic compounds and their nomenclature.	Selection rules, line width and broadening.	Selection rules, line width and broadening.	Selection rules, line width and broadening.	Selection rules, line width and broadening.	Selection rules, line width and broadening.
1	3	3	3	3	3
	CO2	CO2	CO2	CO2	CO2
	Various spectroscopic techniques.	Various spectroscopic techniques.	Various spectroscopic techniques.	Various spectroscopic techniques.	Various spectroscopic techniques.
	1	1	1	1	1
	CO3	CO3	CO3	CO3	CO3
	Importance of spectroscopy for structural elucidation.	Importance of spectroscopy for structural elucidation.	Importance of spectroscopy for structural elucidation.	Importance of spectroscopy for structural elucidation.	Importance of spectroscopy for structural elucidation.
	3	3	3	3	3
	PO1	PO1	PO1	PO1	PO1
	PO2	PO2	PO2	PO2	PO2
	PO3	PO3	PO3	PO3	PO3
	PO4	PO4	PO4	PO4	PO4
	PO5	PO5	PO5	PO5	PO5
	PO6	PO6	PO6	PO6	PO6
	PO7	PO7	PO7	PO7	PO7
	PO8	PO8	PO8	PO8	PO8
	1	3	3	3	3
	CO1	CO1	CO1	CO1	CO1
	The students will acquire knowledge of Distillation and separation	The students will acquire knowledge of Distillation and separation	The students will acquire knowledge of Distillation and separation	The students will acquire knowledge of Distillation and separation	The students will acquire knowledge of Distillation and separation
	1	1	1	1	1
	CO2	CO2	CO2	CO2	CO2
	The students will acquire knowledge of Different chromatographic techniques	The students will acquire knowledge of Different chromatographic techniques	The students will acquire knowledge of Different chromatographic techniques	The students will acquire knowledge of Different chromatographic techniques	The students will acquire knowledge of Different chromatographic techniques
	1	1	1	1	1
	CO3	CO3	CO3	CO3	CO3
	The students will acquire knowledge of Syntheses of various organic compounds and their structural analysis	The students will acquire knowledge of Syntheses of various organic compounds and their structural analysis	The students will acquire knowledge of Syntheses of various organic compounds and their structural analysis	The students will acquire knowledge of Syntheses of various organic compounds and their structural analysis	The students will acquire knowledge of Syntheses of various organic compounds and their structural analysis
	2	2	2	2	2
	PO1	PO1	PO1	PO1	PO1
	PO2	PO2	PO2	PO2	PO2
	PO3	PO3	PO3	PO3	PO3
	PO4	PO4	PO4	PO4	PO4
	PO5	PO5	PO5	PO5	PO5
	PO6	PO6	PO6	PO6	PO6
	PO7	PO7	PO7	PO7	PO7
	PO8	PO8	PO8	PO8	PO8
	2	2	2	2	2

Subject	S Code	Semester	Credit	Duration (Hrs)	L T P	COs	Statement	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8
Analytical Chemistry	MCHM1-261	2	4	45	4.0.0	CO1	Acquire knowledge of basic concepts and importance of analytical chemistry	2		1		1			
						CO2	Acquire knowledge of significance of significant figures and data analysis	2		1					
						CO3	Acquire knowledge of thermogravimetric, electroanalytical, chromatographic methods of analysis	2		1		1			
						CO4	Acquire knowledge of electron microscopic techniques and their application	2				1			
Bioinorganic chemistry	MCHM1-262	2	4	45	4.0.0	CO1	Structures, properties and transport mechanisms of enzymes in physiological systems			2					
						CO2	Metal complexation with various nucleic acids and their role in transcription of nucleic acids.				3				
								PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8

	Subject	S Code	Semester	Credit	Duration (Hrs)	I T P	COs	Statement	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	
Heterocyclic chemistry	MCHM1-313	3	4	45	4.0.0	CO1	Be familiar with the structures of important classes of heterocyclic aromatic organic compounds,				3						
						CO2	Be able to classify simple heterocyclic aromatic compounds as electron deficient or electron rich and explain their reactivity based on these properties,				3						
						CO3	Be able to explain on a mechanistic level, reactions and synthesis of important electron deficient nitrogen containing heterocycles; pyridines, diazines and their benzo-condensed analogs,					2					
						CO4	Be able to explain on a mechanistic level, reactions and synthesis of important										2
						CO2	Electronic and Hamiltonian operators for molecules							2			
						CO3	Quantum chemical description of angular momentum and term symbols for a one and manyelectron systems							2			
						CO4	Born-Oppenheimer approximation, the Pauli principle, Hund's rules, Hückel theory and the variation principle							2			

						of the spectra (NMR, IR, UV & MS).								
					CO2	The students will acquire knowledge of Various reactions conditions including modern coupling strategies and their implications	1			2			3	

Enter Correction levels 1, 2 or 3 as defined below:

1. Slight (Low) - upto 30% 2. Moderate (Medium) – above 30% and upto70% 3. Substantial (High) – above 70%

So on..... (1st semester to last semester)